Alder Coppice Primary School Knowledge Organiser

Subject: Science

Year: 5

Unit 5: Reversible Changes

Links to: Unit 3 Separating Materials and Y4 Solids, Liquids and Gases

What I Should Already Know:

- That changes can be reversed (Y5)
- About Solids, liquids and gases (Y4)

Skills and Enquiry:

- What causes irreversible changes?
- What causes rusting?
- How can we make new materials from chemical reactions?
- What is the difference between heating and burning?

Fair test—A fair test is a test which controls all but one variable when attempting to answer a scientific question. Only changing one variable allows the person conducting the test to know that no other variable has affected the results of the test.

Unit Specific Vocabulary:

carbon dioxide – a gas made every time something is burnt

chemical reaction- what happens when there is an irreversible change

combustion – the scientific word for burning

irreversible - a change that can only go one way, also known as non-reversible or permanent changes.

product - the new material made as a result of an irreversible change

property - something about how a material feels, appears and is measured i.e. strong, soft

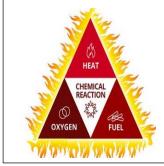
reversible change – a change where a material can be returned to its original state i.e. freezing water and melting ice

rusting - The irreversible change which causes the weakening (corrosion) of iron and steel in the presence of oxygen and water.

yeast - a helpful fungus used by the food industry

What I should know by the end of this Unit:

- That some changes result in the formation of new materials, and that this kind of change is not usually reversible, including changes associated with burning and the action of acid on bicarbonate of soda.
- That heating and burning are two different things
- How a famous chemist has created a new material.



Key Facts:

What are irreversible changes? Irreversible changes are changes that cannot be undone or reversed.

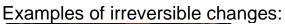








Heat







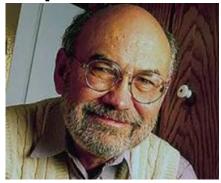








Significant Scientists – chemists **Spencer Silver**



Ruth Benrito

